Agelastatin E, Agelastatin F, and Benzosceptrin C from the Marine Sponge Agelas dendromorpha

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The study of the *n*-butanol extract of the New Caledonian sponge *Agelas dendromorpha* led to the isolation and identification of three new pyrrole-2-aminoimidazole (P-2-AI) alkaloids, named agelastatins E (**3**) and F (**4**) and benzosceptrin C (**5**), together with 10 known metabolites, agelastatin A (**1**), agelastatin D (**2**), sceptrin (**6**), manzacidin A, tauroacidin A, taurodispacamide A, nortopsentin D, thymine, longamide, and 4,5-dibromopyrrole-2-carboxamide. Their structures were assigned by spectroscopic data interpretation. All the compounds were tested for cytotoxic activity.

Marine alkaloids possessing a pyrrole-2-imidazole (P-2-AI) structure are known to be some of the most common and structurally intriguing metabolites isolated from marine sponges. Over the last 30 years, a number of molecules have been encountered that likely arise from clathrodin,¹ hymenidin,² and oroidin precursors.³ Many of these compounds are reported to have interesting biochemical activities, such as blocking α -adrenoceptors,⁴ serving as an antagonist of serotonergic receptors,5 activating actomyosin AT-Pase,⁶ and inhibiting kinase activity,⁷ as well as antibacterial,⁸ antifungal,8 and antihistamine activities.9 Among them, the immunosuppressive activity of palau'amine,¹⁰ the kinase inhibition of debromohymenialdesin,¹¹ the antitumor activity of girolline,¹² and the cytotoxicity of agelastatin A¹³ are remarkable. Agelastatin A (1), along with a minute quantity of agelastatin B, was first reported from the sponge Agelas dendromorpha.^{13a} Later on, in addition to agelastatins A and B, the closely related compounds agelastatins C and D were isolated from the sponge Cymbastela sp.^{13b}

During our search for bioactive substances from marine organisms, we have previously reported several pyrrole-2-aminoimidazole (P-2-AI) alkaloids, together with a proposal for their biogenesis and biomimetic synthesis.¹⁴ In line with the pursuit of biological activities and the structure-guided isolation for biosynthetic investigations, we have recently carried out a chemical study of the New Caledonian sponge Agelas dendromorpha (Lévi, 1993) (order Agelasida, family Agelasidae). The freeze-dried sponge was extracted with MeOH, and the extract was partitioned between n-BuOH and H₂O. The n-BuOH fraction was subjected to silica gel column chromatography followed by preparative RP₁₈ HPLC to yield three new minor compounds, agelastatin E (3), agelastatin F (4), and benzosceptrin C (5) together with a series of known alkaloids. These were agelastatin A (1),^{13a} agelastatin D (2),^{13b} sceptrin (**6**),¹⁵ manzacidin A,¹⁶ tauroacidin A,¹⁷ taurodispacamide A,¹⁸ nortopsentin D,¹⁹ thymine, longamide,²⁰ and 4,5-dibromopyrrole-2-carboxamide.²¹ The structures of the known metabolites were determined by comparison of their ¹H and ¹³C NMR spectra with those reported in the literature. Here, we describe the isolation of the new compounds 3, 4, and 5.

The ESI mass spectrum of agelastatin E (**3**) showed pseudomolecular ion peaks at m/z 353:355 (1:1) [M + H]⁺, indicating the presence of one bromine atom in the molecule. The molecular formula of **3** was revealed as C₁₃H₁₅⁷⁹BrN₄O₃ by HRESIMS. The



UV absorption at λ_{max} 286 nm was attributed to a substituted pyrrole chromophore, while IR absorption at 3223, 1656, and 1642 cm⁻¹ indicated the occurrence of NH and amide carbonyl functionalities. The ¹H and ¹³C NMR spectra of compound **3** were similar to those of agelastatin A (1),^{13a} except for an additional signal arising at $\delta_{\rm H}$ 3.18 (s, 3H) for either an $-\rm OCH_3$ or $-\rm NCH_3$ group (Table 1). The position of the CH₃ with $\delta_{\rm H}$ 3.18 ($\delta_{\rm C}$ 50.8) was established as an $-OCH_3$ at C-5 on the basis of its HMBC correlation with δ_C 101.0 (C-5). The signal at $\delta_{\rm H}$ 2.78 (3H, s) showed correlations with amide carbonyls at C-2 ($\delta_{\rm C}$ 162.2) and 101.0 (C-5), confirming the presence of an NCH₃ group (Figure 1). The ¹H-¹H COSY spectrum showed the connectivities from H-6a/b ($\delta_{\rm H}$ 2.66, dd, J =12.9, 6.6 Hz/2.14, dd, J = 12.9 Hz) to H-7 ($\delta_{\rm H}$ 4.62, dd, J = 12.6, 6.6 Hz); H-7 to H-8 ($\delta_{\rm H}$ 4.11, d, J = 5.4 Hz); and H-14 ($\delta_{\rm H}$ 6.32, d, J = 4.1 Hz) to H-15 ($\delta_{\rm H}$ 6.91, d, J = 4.1 Hz) for a cyclopentane ring and a monobromopyrrole moiety, respectively. The pyrrole signals at $\delta_{\rm H}$ 6.32 (d, J = 4.1 Hz, H-14) and $\delta_{\rm H}$ 6.91 (d, J = 4.1Hz, H-15) showed HMBC correlations at $\delta_{\rm C}$ 124.2 (C-11), 107.5 (C-13), 116.2 (C-15) and $\delta_{\rm C}$ 124.2 (C-11), 107.4 (C-13), 114.0 (C-

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Table 1. ¹H and ¹³C NMR Data for Agelastatins E (3) and F (4) Recorded in CD₃OD (600 MHz)

	agelastatin E (3)				agelastatin F (4)			
position	$\delta_{\rm C}$	$\delta_{\mathrm{H}} \left(J \text{ in Hz} \right)$	COSY	HMBC ^b	$\delta_{\rm C}$	$\delta_{\mathrm{H}} \left(J \text{ in Hz} \right)$	COSY	HMBC ^b
1-NH								
2	162.2				162.8			
3-NH		7.3				7.3		
4	61.2	4.08, bs		2, 5, 7, 8	69.5	3.92, bs		2, 5, 7
5	101.0				93.3			
6a	39.3	2.66, dd (12.9, 6.6)	H-6b, H-7	5, 6a, 7	43.7	2.58, dd (12.9, 6.6)	H-6b, H-7	4, 5, 6, 7, 8
6b		2.14, bt (12.9)	H-6a, H-7	5, 6a, 7, 8		2.24, bt (12.9)	H-6a, H-7	5, 6, 7
7	53.9	4.62, m (12.6, 6.6)	H-8	6, 11, 13	55.8	4.74, m (12.6, 6.6)	H-8	6, 7, 11, 13
8	62.2	4.11, d (5.4)	H-7	4, 6	62.2	4.14, d (5.5)	H-7	4, 5, 6
9-NH		7.92				8.1		
10	161.2				160.2			
11	124.2				125.0			
12								
13	107.4				108.7			
14	114.0	6.32, d (4.1)	H-15	11, 13, 15	101.1			
15	116.2	6.91, d (4.1)	H-14	11, 13, 14	117.1	6.98, s		11, 14
N-Me	24.7	2.78, s		2, 5, 6				
O-Me	50.8	3.18, s		5				

^{*a*} Shifts of D₂O exchangeable protons were detected in DMSO- d_6 at 500 MHz: δ_H 6.86, 6.39 for agelastatins E (3) and F (4); δ_H 5.7, 8.04, 8.34 for benzosceptrin C (5). ^{*b*} HMBC correlations are from the proton stated to the indicated carbon.



Figure 1. Key COSY, HMBC, and NOESY correlations of agelastatins E(3) and F(4) and benzosceptrin C (5).

14), respectively. The HMBC correlations of $\delta_{\rm H}$ 4.62 (H-7) with $\delta_{\rm C}$ 124.3 (C-11) and 107.4 (C-13) established the connection between the cyclopentane ring and the pyrrole moiety. Further analysis of the HMBC experiment established the connectivity of H-6 methylene protons at $\delta_{\rm H}$ 2.66 (dd) and 2.14 (br t) to C-5, C-6, C-7 and C-5, C-6, C-7, C-8, respectively. The broad singlet at $\delta_{\rm H}$ 4.08 (H-4) showed correlations with C-2, C-5, C-7, and C-8, while the doublet at $\delta_{\rm H}$ 4.11 (H-8) showed correlations with C-4 and C-6. The quaternary carbon signals $\delta_{\rm C}$ 162.2 (C-2), 101.0 (C-5), 124.3 (C-11), and 107.4 (C-13), invisible in the 1D ¹³C NMR spectrum, have been confirmed by HMBC experiment. Taken together, the data and the very similar 1D ¹H NMR spectrum to that of agelastatin A (1) gave the structure 3 for this compound, which was named agelastatin E. NOESY experiments (Figure 1) showed correlations between the OMe group at position C-5 and H-4 on one hand and between H-7 and H-6/H-8 on the other hand, indicating the cisfused cycles at C_4-C_5 and C_7-C_8 . The relative configuration for 3 was revealed to be the same as for agelastatin A (1). Whether agelastatin E (3) is a naturally methylated derivative of agelastatin A (1) or just a result of its dehydration and methoxylation is an interesting question. The second process could occur during the purification process in the presence of MeOH.

Agelastatin F (4) was obtained as a white solid and was sparingly soluble in MeOH. The positive ESI-MS spectrum showed pseudomolecular ion peaks at m/z 405, 407, and 409 (1:2:1) for the presence of two bromine atoms in the molecule. The molecular formula of **4** was revealed as $C_{11}H_{10}^{79}Br_2N_4O_3$ by (-)-HRESIMS. The ¹H NMR spectrum of **4** was very similar to that of agelastatin D (2),^{13b} but lacked a proton signal at $\delta_{\rm H}$ 6.33 (H-14), which is replaced by a Br atom, as revealed by the molecular formula. The pyrrole signal at $\delta_{\rm H}$ 6.98 showed HMBC correlations to the carbons at $\delta_{\rm C}$ 125.03 (C-11), 108.7 (C-13), and 101.1 (C-14) (Table 1). There were HMBC correlations from H-7 to the pyrrolic C-11 and C-13. The H-15 ($\delta_{\rm H}$ 6.98) bonded to C-15 ($\delta_{\rm C}$ 117.1) showed strong HMBC correlations with C-11 and C-14, allowing for only the 13,14 bromine substitution pattern, positioning the hydrogen at C-15. The strongly down-shifted C-14 (δ_{C} 101.1) and the comparison with NMR spectra published for agelastatin B^{13b,e} indicated that this position was brominated. The UV absorption at λ_{max} 288 nm was attributed to a substituted pyrrole chromophore. The IR spectrum showed absorption bands at 3254, 1681, and 1634 cm⁻¹ for OH, NH, and amide carbonyl groups. All of the connections were established by HMBC and COSY experiments. The ¹H-¹H COSY spectrum showed the connectivities from H-6a/b to H-7 and H-7 to H-8 (Figure 1). The methylene protons (H-6) of the cyclopentane ring, $\delta_{\rm H}$ 2.24 (br t) and 2.58 (dd), showed HMBC connections with $\delta_{\rm C}$ 93.3 (C-5), 43.7 (C-6), 55.8 (C-7) and $\delta_{\rm C}$ 69.5 (C-4), 93.3 (C-5), 43.7 (C-6), 55.8 (C-7), 62.2 (C-8), respectively. The singlet at $\delta_{\rm H}$ 3.92 was connected to 162.8 (C-2), 93.3 (C-5), and 55.8 (C-7). The multiplet at $\delta_{\rm H}$ 4.74 (H-7) and doublet at $\delta_{\rm H}$ 4.14 (H-8) showed correlations with $\delta_{\rm C}$ 93.3 (C-5) and $\delta_{\rm C}$ 43.7 (C-6), 125.0 (C-11), 108.7 (C-14), respectively. As for compound 3, NOESY experiments of 4 showed clear correlations between H-7 and H-6/H-8, while H-4 does not correlate with any proton. These observations together with the very similar 1D ¹H NMR spectra of 3 and 4 included in the Supporting Information indicate that the relative configuration of 4 must be the same as for 3.

The ESI-MS of compound **5** displayed five pseudomolecular ion peaks at m/z 771, 773, 775, 777, and 779 $[M + H]^+$, which indicated the presence of four bromines in the molecule. The molecular formula of **5** was determined to be $C_{22}H_{18}^{79}Br_4N_{10}O_2$ by (–)-HRESIMS. The ¹H NMR spectrum displayed only a few protons at δ_H 3.43(m)/3.97(dd) for H-8, δ_H 3.71(dd, H-7), and δ_H 6.60(d, H-15), suggesting it to be a symmetric molecule. The comparison of the ¹H NMR spectrum of **5** with the data of benzosceptrins A (7) and B (8), recently isolated in our group,²² suggested it to be a benzosceptrin-type compound with the benzocyclobutane ring system. The ¹H NMR spectra were similar in the upfield region with the difference in the pyrrole proton region. Only a single pyrrole proton peak at $\delta_{\rm H}$ 6.60 (2H, H-15) was observed. Thus, the molecule differs from benzosceptrins A (7) and B (8) in the degree of bromination. Interestingly, the chemical shifts of the remaining pyrrolic protons are not dramatically changed with the degree of bromination. Comparison of the chemical shifts of H-13 ($\delta_{\rm H}$ 8.86), H-14 ($\delta_{\rm H}$ 6.08), and H-15 ($\delta_{\rm H}$ 6.61) in benzosceptrin A $(7)^{22}$ with H-13 ($\delta_{\rm H}$ 8.86) and H-15 ($\delta_{\rm H}$ 6.61) in the dibrominated benzosceptrin B (8) and the signal at $\delta_{\rm H}$ 6.60 in 5 clearly show that it corresponds to H-15. Similar chemical shifts have been observed for debromosceptrin,23 sceptrin, and dibromosceptrin.²⁴ The ¹³C NMR spectrum gave the characteristic values for the brominated carbons at $\delta_{\rm C}$ 100.3 (C-14) and 106.4 (C-13). The intense signal at $\delta_{\rm C}$ 114.4 was attributed C-15. This was confirmed by the HMBC correlations of H-15 with C-10 at $\delta_{\rm C}$ 162.1 and C-11 at $\delta_{\rm C}$ 128.6. In fact, the HMBC correlations were similar to those observed for benzosceptrin A (Figure 1).²² The ¹H-¹H COSY spectrum showed correlations between $\delta_{\rm H}$ 3.43 (m, H-8a), 3.97 (dd, H-7), and 3.71 (dd, H-8b). The ¹³C NMR spectrum gave the corresponding carbon values at δ_{C} 49.7 (CH-7) and 43.7 (CH₂-8).

The specific rotation, $[\alpha]^{25}_{\rm D} - 22.7$, showed the C7–C7' *trans* configuration, indicating the preservation of the C_2 symmetry axis like for sceptrin (**6**) and benzosceptrins A and B. A *cis* configuration for C7–C7' would give the achiral *meso* compound. This third remarkable and unique benzocyclobutane metabolite was named benzosceptrin C (**5**).²⁵

For biogenetic considerations, we have previously suggested that the agelastatin skeleton could be derived from oroidin through subtle tautomerization and the dual reactivity induced by the 2-aminoimidazole portion of the structure.^{14a} A preagelastatin motif like that we have proposed is clearly shown by the dimeric natural product nagelamide J, recently isolated by Kobayashi and co-workers.²⁶ A discussion regarding the mechanism for formation of the benzosceptrin skeleton has also been reported in our previous paper.²²

The pure compounds were screened for cytotoxicity against the KB cell line,²⁷ and except for agelastatin A (1) (100% activity at 30 and 3 μ M), all of the compounds lacked significant bioactivity at 30 μ M. The pharmacological activities of agelastain A have been extensively studied.²⁸ The sharp difference between the high activity of agelastatin A compared to the natural derivatives agelastatins B–F reveals very little flexibility for synthetic analogues and structure–activity relationship studies.

In conclusion, the majority of the compounds isolated from *A. dendromorpha* are of the pyrrole-2-aminoimidazole type. In addition to the known cytotoxic compounds agelastatin A (1) and agelastatin D (2), which have already been isolated from the same sponge, we have isolated agelastatins E (3) and F (4). Benzosceptrin C (5) was also isolated as a new P-2-AI derivative. Moreover, metabolites having a substituted benzocyclobutane such as benzosceptrin C (5), described in this paper, and the benzosceptrins A and B,²² which we have reported earlier, are completely new. Their significance for chemistry, biology, and chemical ecology is an interesting question. Accordingly, the presence of these compounds should stimulate considerable synthetic interest. It is noteworthy that there are very few examples of benzocyclobutane-containing natural products.

Experimental Section

General Experimental Procedures. Optical rotations were measured in MeOH on a Jasco P1010 polarimeter. The UV spectra were recorded on a Waters 996 photodiode array detector in the HPLC solvent. The IR spectra were measured (neat) on a Perkin-Elmer BX-FT-IR spectrometer. NMR spectra were recorded on Bruker Avance 600 MHz and DRX 500 MHz spectrometers. The chemical shifts were referenced to the residual solvent signal (CD₃OD, $\delta_{\rm H} = 3.32$, $\delta_{\rm C} = 49.0$). HRMS data were obtained with a hybrid linear trap/orbitrap mass spectrometer (LTQ-orbitrap, ThermoFisher) in electrospray ionization mode by direct infusion of the purified compounds. Preparative HPLC was performed on an autoprep system (Waters 600 controller and Waters 600 pump with a Waters 996 photodiode array detector). Samples were injected by a Waters 2700 sample manager.

Sponge Material. The sponge *Agelas dendromorpha* (order Agelasida, family Agelasidae) was collected during the research cruise Terrasses (October 2008) on a sea mount of the Norfolk rise (23°41′ S, 168°00′ E), New Caledonia, at a depth of 250 m. The voucher sample is preserved in Muséum National d'Histoire Naturelle in Paris under the accession number ORSTOM-R1569.

Extraction and Isolation. The lyophilized sponge A. dendromorpha (150 g) was extracted with MeOH (500 mL \times 3). After removing the solvent in vacuo, the combined MeOH extract was partitioned between *n*-BuOH (200 mL \times 3) and H₂O to afford an *n*-BuOH fraction (6.5 g). This n-BuOH fraction was chromatographed on a silica gel column (VersaPak, Sigma, 40-150 mm) with gradient elution of MeOH-CH₂Cl₂ to give 15 subfractions (B1-B15). The alkaloid-rich fractions B8-B15 were further purified on a preparative C18 HPLC (X-bridge column, 19-150 mm), flow rate 17 mL/min, MeOH-0.1% HCOOH and H₂O-0.1% HCOOH, UV detection at 270 nm. Further purification of the B8 fraction yielded thymine (1.6 mg, t_R 12.2 min), longamide (5.6 mg, t_R 33.1 min), and 4,5-dibromopyrrole-2-carboxamide (6.5 mg, t_R 35.7 min). Fractions B10 and B11 yielded a major compound, agelastatin A (1, 380 mg, t_R 31.0 min). The purification of fractions B12-15 afforded three new compounds along with several known compounds, agelastatin D (2, 2.0 mg, t_R 26.1 min), agelastatin E (3, 1.5 mg, t_R 34.4 min), agelastatin F (4, 20.9 mg, t_R 34.1 min), benzosceptrin C (5, 0.6 mg, t_R 35.1 min), sceptrin (6, 9.3 mg, t_R 17.8 min), manzacidin A (16.0 mg, t_R 29.5 min), tauroacidin A (16.9 mg, t_R 32.7 min), taurodispacamide (37.2 mg, t_R 32.7 min), and nortopsentin D (34.1 mg, t_R 35.7 min). The HPLC (LC-MS) of all the isolated compounds was carried out on an analytical column (Sunfire C18, 5 μ m, 4.6–150 mm, flow rate 0.7 mL/min) with an isocratic gradient for 5 min using 98:2 (H₂O-0.1% HCOOH to MeOH-0.1% HCOOH) followed by a linear gradient to 100% MeOH-0.1% HCOOH in 45 min. The retention time, $t_{\rm R}$, for all the isolated compounds is based on this analytical HPLC.

Agelastatin A (1): white solid; $[\alpha]^{25}_{D}$ = 58.5 (*c* 0.21, MeOH) (lit.^{13a} = 84.3 (*c* 0.3, EtOH)).

Agelastatin D (2): white solid, $[\alpha]^{25}_{D}$ -12 (*c* 0.07, MeOH) (lit.^{13b} not given).

Agelastatin E (3): white solid; $[α]^{25}_{D} -28$ (*c* 0.09, MeOH); UV (MeOH) $λ_{max}$ (log ε) 232 (3.95), 286 (4.08); IR (neat) $ν_{max}$ 3223, 2933, 1656, 1642, 1552, 1421, 1351, 1197, 1082, 1037 cm⁻¹; ¹H and ¹³C NMR (CD₃OD, 600 MHz), Table 1; ESIMS *m*/*z* 355, 357 (1:1) [M + H]⁺; HRESIMS *m*/*z* 353.0276 [M - H]⁻ (calcd for C₁₃H₁₅⁷⁹BrN₄O₃, *m*/*z* 353.0255).

Agelastatin F (4): colorless, amorphous solid; $[α]^{25}{}_D - 34.3$ (*c* 0.11, MeOH); UV (MeOH) $λ_{max}$ (log ε) 208 (4.0), 238 (3.98), 288 (4.18); IR (neat) $ν_{max}$ 3254, 1681, 1634, 1552, 1373, 1346, 1221, 1112, 1067 cm⁻¹; ¹H and ¹³C NMR (CD₃OD, 600 MHz), Table 1; ESIMS *m/z* 405, 407, 409 [M + H]⁺; HRESIMS *m/z* 402.9071 [M - H]⁻ (calcd for C₁₁H₁₀⁷⁹Br₂N₄O₃, *m/z* 402.9048).

Benzosceptrin C (5): white solid; $[α]^{25}_{D} - 22.7$ (*c* 0.13, MeOH); UV (MeOH) λ_{max} (log ε) 229 (4.04), 276 (3.90); IR (neat) ν_{max} 3156, 1679, 1577, 1565, 1555, 1347, 1228, 1096, 1038, 750, 613 cm⁻¹; ¹H NMR (CD₃OD, 600 MHz) 3.71 (dd, J = 6.7, 3.8 Hz), 3.97 (dd, J =10.1, 3.9 Hz), 3.43 (m), 6.60 (s); ¹³C NMR (CD₃OD, 150.8 MHz) 43.7 (C8, C8'), 49.7 (C7, C7'), 100.3 (C14, C14'), 106.4 (C13, C13'), 114.4 (C15, C15'), 120.2 (C6, C6'), 128.6 (C11, C11'), 154.5 (C2, C2'), 162.1 (C10, C10'), C4 and C5 were not detected; ESI-MS *m/z* 771, 773, 775, 777, 779; (-)-HRESIMS *m/z* 768.8301 [M – H]⁻ (calcd for C₂₂H₁₈⁷⁹Br₄N₁₀O₂, *m/z* 768.8277).

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References and Notes

- (1) Morales, J. J.; Rodriguez, A. D. J. Nat. Prod. 1991, 54, 629-631.
- (2) Kobayashi, J.; Ohizumi, Y.; Nnakamura, H.; Hirata, Y. *Experientia* 1986, 42, 1176–1177.
- (3) Forenza, S.; Minale, L.; Riccio, R.; Fattorusso, E. J. Chem. Soc., Chem. Commun. 1971, 1129–1130.
- (4) Kobayashi, J.; Nakamura, H.; Ohizumi, Y. *Experientia* 1988, 86–87.
 (5) Cafieri, F.; Fattorusso, E.; Mangoni, A.; Tagliatatela-Scafati, O.;
- Carnuccio, R. Bioorg. Med. Chem. Lett. 1995, 8, 799–804.
- (6) Kobayashi, J.; Tsuda, M.; Ohizumi, Y. Experientia 1991, 47, 301– 304.
- (7) Meijer, L.; Thumissen, A. M. W. H.; White, A. W.; Garnier, M.; Nikolic, M.; Tsai, L. H.; Walter, J.; Cleverley, K. E.; Salinas, P. C.; Wu, Y. Z.; Biernat, J.; Mandelkow, E. M.; Kim, S. H.; Pettit, G. R. *Chem. Biol.* **2000**, *7*, 51–63.
- (8) Cafieri, F.; Fattorusso, E.; Tagliatatela-Scafati, O. J. Nat. Prod. 1998, 61, 122–125.
- (9) Cafieri, F.; Carnuccio, R.; Fattorusso, E.; Tagliatatela-Scafati, O.; Vallefuoco, T. *Bioorg. Med. Chem. Lett.* **1997**, *7*, 2283–2288.
- (10) (a) Kinnel, R. B.; Gehrken, H. P.; Scheuer, P. J. J. Am. Chem. Soc. 1993, 115, 3376–3377. (b) Kato, T.; Shizuri, Y.; Izumida, H.; Yokoyama, A.; Endo, M. Tetrahedron Lett. 1995, 36, 2133–2136. (c) Overman, L. E.; Rogers, B. N.; Tellew, J. E.; Trenkle, W. C. J. Am. Chem. Soc. 1997, 119, 7159–7160. (d) Kinnel, R. B.; Gehrken, H. P.; Swali, R.; Skoropowski, G.; Scheuer, P. J. J. Org. Chem. 1998, 63, 3281–3286. (e) Jacquot, D. E. N.; Lindel, T. Curr. Org. Chem. 2005, 9, 1551–1565.
- (11) Tasdemir, D.; Mallon, R.; Greenstein, M.; Feldberg, L. R.; Kim, S. C.; Collins, K.; Wojciechowicz, D.; Mangalindan, G. C.; Concepción, G. P.; Harper, M. K.; Ireland, C. M. J. Med. Chem. 2002, 45, 529– 532.
- (12) Lavelle, F.; Zerial, A.; Fizames, C.; Rabault, B.; Curaudeau, A. Invest. New Drugs 1991, 9, 233–244.
- (13) (a) D'Ambrosio, M.; Guerriero, A.; Debitus, C.; Ribes, O.; Pusset, J.; Leroy, S.; Pietra, F. J. Chem. Soc., Chem. Commun. 1993, 1305– 1306. (b) Hong, T. W.; Jı'menez, D. R.; Molinski, T. F. J. Nat. Prod. 1998, 61, 158–161. (c) D'Ambrosio, M.; Guerriero, A.; Ripamonti, M.; Debitus, C.; Waikedre, J.; Pietra, F. Helv. Chim. Acta 1996, 79, 727–735. (d) Mason, C. K.; McFarlane, S.; Johnston, P. G.; Crowe, P.; Erwin, P. J.; Domostoj, M. M.; Campbell, F. C.; Manaviazar, Hale, K. J.; El-Tanani, M. Mol. Cancer Ther. 2008, 7, 548–558. (e) Feldman, K. S.; Saunders, J. C. J. Am. Chem. Soc. 2002, 124, 9060–9061.
- (14) (a) Al-Mourabit, A.; Potier, P. *Eur. J. Org. Chem.* 2001, 237–243.
 (b) Travert, N.; Martin, M.-T.; Bourguet-Kondracki, M. L.; Al-Mourabit, A. *Tetrahedron Lett.* 2005, 46, 249–252. (c) Vergne, C.; Bourg-Esnault, N.; Perez, T.; Martin, M. T.; Adeline, M.-T.; Dau, E. T. H.; Al-Mourabit, A. *Org. Lett.* 2006, 8, 2421–2424. (d) Schroif-

Gregoire, C.; Travert, N.; Zaparucha, A.; Al-Mourabit, A. *Org. Lett.* **2006**, *8*, 2961–2964. (e) Vergne, C.; Appenzeller, J.; Ratinaud, C.; Martin, M.-T.; Debitus, C.; Zaparucha, A.; Al-Mourabit, A. *Org. Lett.* **2008**, *10*, 493–496. (f) Picon, S.; Dau, E. T. H.; Martin, M.-T.; Retailleau, P.; Zaparucha, A.; Al-Mourabit, A. *Org. Lett.* **2009**, *11*, 2525–2526.

- (15) (a) Walker, R. P.; Faulkner, D. J.; Van Engen, D.; Clardy, J. J. Am. Chem. Soc. 1981, 103, 6772–6773. (b) Cafieri, F.; Carnuccio, R.; Fattorusso, E.; Taglialatela-Scafati, O.; Vallefuoco, T. Biorg. Med. Chem. Lett. 1997, 7, 2283–2288. (c) Shen, X.; Perry, P. L.; Dunbar, C. D.; Kelly-Borges, M.; Hamann, M. T. J. Nat. Prod. 1988, 61, 1302– 1303.
- (16) Kobayashi, J.; Kanda, F.; Ishibashi, M.; Shigemori, H. J. Org. Chem. 1991, 56, 4574–4576.
- (17) Kobayashi, J.; Inaba, K.; Tsuda, M. Tetrahedron 1997, 53, 16679– 16682.
- (18) Fattorusso, E.; Taglialatela-Scafati, O. *Tetrahedron Lett.* **2000**, *41*, 9917–9922.
- (19) Mancini, I.; Guella, G.; Pietra, F.; Debitus, C.; Waikedre, J. *Helv. Chim. Acta* **1996**, *79*, 2075–2082.
- (20) (a) Cafieri, F.; Fattorusso, E.; Mangoni, A.; Taglialatela-Scafati, O. *Tetrahedron Lett.* **1995**, *36*, 7893–7896. (b) Umeyama, A.; Ito, S.; Yuasa, E.; Arihara, S.; Yamada, T. J. Nat. Prod. **1998**, *61*, 1433–1434.
- (21) Nakamura, H.; Wu, H.; Kobayashi, J.; Kobayashi, M.; Ohizumi, Y.; Hirata, Y. J. Org. Chem. 1985, 50, 2494–2497.
- (22) Appenzeller, J.; Tilvi, S.; Gallard, J.-F.; Laurent, D.; Debitus, C.; Moriou, C.; Al-Mourabit, A. Org. Lett. 2009, 11, 4874–4877.
- (23) Shen, X.; Perry, T. L.; Dunbar, C. D.; Kelly-Borges, V.; Hamann, M. T. J. Nat. Prod. 1998, 61, 1302–1303.
- (24) Birman, V. B.; Jiang, X.-T. Org. Lett. 2004, 6, 2369-2371.
- (25) During the review process for this manuscript, the structure of benzosceptrin C was independently reported: Kubota, T.; Araki, A.; Yasuda, T.; Tsuda, M.; Fromont, J.; Aoyama, K.; Mikami, Y.; Wälchli, M. R.; Kobayashi, J. *Tetrahedron Lett.* **2009**, *50*, 7268–7270.
- (26) Araki, A.; Tsuda, M.; Kubota, T.; Mikami, Y.; Fromont, J.; Kobayashi, J. Org. Lett. 2007, 9, 2369–2371.
- (27) Zampella, A.; Sepe, V.; Bellotta, F.; Luciano, P.; d'Auria, M. V.; Cresteil, T.; Debitus, C.; Petek, S.; Poupat, C.; Ahond, A. *Org. Biomol. Chem.* **2009**, *7*, 4037–4044.
- (28) (a) Pettit, G. R.; Ducki, S.; Herald, D. L.; Doubek, D. L.; Schmidt, J. M.; Chapuis, J. C. Oncol. Res. 2005, 15, 11–20. (b) Hale, K. J.; Domostoj, M. M.; El-Tanani, M.; Campbell, F. C.; Mason, C. K., Eds. In Strategies and Tactics in Organic Synthesis; Academic Press: London, 2005; pp 352–394. (c) Mason, C. K.; McFarlane, S.; Johnston, P. G.; Crowe, P.; Erwin, P. J.; Domostoj, M. M.; Campbell, F. C.; Manaviazar, S.; Hale, K. J.; El-Tanani, M. Mol. Cancer Ther. 2008, 7, 548–558.
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